

### 3M™ NOVEC™ 1230 FIRE PROTECTION FLUID

#### Description

Novec 1230 fire protection fluid is a colorless, clear liquid (see Physical Properties Table for additional information). It is stored as a liquid and dispensed into the hazard as an electrically non-conductive gaseous vapor that is clear and does not obscure vision. It leaves no residue and has acceptable toxicity for use in occupied spaces at design concentration.

#### Extinguishing Method

Novec 1230 fire protection fluid extinguishes a fire by heat absorption. The gaseous mixture created when Novec 1230 fluid discharges into air has a much higher heat capacity than air alone. The gaseous mixture absorbs large amounts of heat due to the high heat capacity and extinguishes fires by sufficiently cooling the combustion zone. It is important to note, Novec 1230 fluid does not use the depletion or displacement of oxygen to extinguish a fire and therefore is safe for use in occupied spaces.

#### Approvals

- Underwriters Laboratories (UL)
- Underwriters Laboratories of Canada (ULC)
- Factory Mutual (FM)
- United States EPA Significant New Alternative Policy (SNAP) report

*For exact certification listings, please reference the respective agency web site.*

#### Use and Limitations

Novec 1230 fire protection fluid can be used on the following class of hazards:

- Class A & C:** Electrical and Electronic Hazards  
Telecommunications Facilities  
High value assets, where the associated down-time would be costly
- Class B:** Flammable liquids and gases

Novec 1230 fire protection fluid shall **NOT** be used on fires involving the following materials:

- Chemicals or mixtures of chemicals that are capable of rapid oxidation in the absence of air such as Cellulose Nitrate and Gunpowder
- Reactive metals such as Lithium, Sodium, Potassium, Magnesium, Titanium, Zirconium, Uranium, and Plutonium
- Metal hydrides such as Sodium Hydride and Lithium Aluminum Hydride
- Chemicals capable of undergoing auto-thermal decomposition such as Organic Peroxides and Hydrazine

#### Exposure Limitations

The discharge of clean agent systems to extinguish a fire can result in potential hazard to personnel from the natural form of the clean agent or from the products of combustion that result from exposure of the agent to the fire or hot surfaces. Unnecessary exposure of personnel either to the natural agent or to the products of decomposition shall be avoided.

| Hazard Type             | Design Concentration | Maximum Human Exposure Time |
|-------------------------|----------------------|-----------------------------|
| Normally Occupied Space | 4.5% to 10%          | 5 minutes                   |

Fike does not recommend systems using Novec 1230 fire protection fluid to be used in any normally occupied spaces where the design concentration required is above 10%

This document is only intended to be a guideline and is not applicable to all situations. Information is subject to Fike's full disclaimer at <http://www.fike.com/disclaimer>.

## Physical Properties

Not for specification purposes. All values @ 25°C (77°F) unless otherwise specified.

|                                                            |                                                                       |
|------------------------------------------------------------|-----------------------------------------------------------------------|
| Chemical Name                                              | Dodecafluoro-2-methylpentan-3-one                                     |
| Chemical Formula                                           | CF <sub>3</sub> CF <sub>2</sub> C(O)CF(CF <sub>3</sub> ) <sub>2</sub> |
| Molecular Wt.                                              | 316.04                                                                |
| Boiling Point @ 1 atm (760 mmHg)                           | 49.2°C (120.6°F)                                                      |
| Freezing Point                                             | -108.0°C (-162.4°F)                                                   |
| Critical Temperature                                       | 168.7°C (335.6°F)                                                     |
| Critical Pressure                                          | 18.65 bar (270.44 psia)                                               |
| Critical Volume                                            | 494.5 cc/mole (0.0251 ft <sup>3</sup> /lbm)                           |
| Critical Density                                           | 639.1 kg/m <sup>3</sup> (39.91 lb/ft <sup>3</sup> )                   |
| Density, Sat. Liquid                                       | 1.60 g/ml (99.9 lb/ft <sup>3</sup> )                                  |
| Density, Gas @ 1 atm                                       | 0.0136 g/ml (0.851 lb/ft <sup>3</sup> )                               |
| Specific Volume, Gas @ 1 atm                               | 0.0733 m <sup>3</sup> /kg (1.175 ft <sup>3</sup> /lb)                 |
| Specific Heat, Liquid                                      | 1.103 kJ/kg-°C (0.2634 Btu/lb-°F)                                     |
| Specific Heat, Vapor @ 1 ATM                               | 0.891 kJ/kg-°C (0.2127 Btu/lb-°F)                                     |
| Heat of Vaporization @ Boiling Point                       | 88.0 kJ/kg (37.9 Btu/lb)                                              |
| Liquid Viscosity @ 0°C/25°C                                | 0.56/0.39 centistokes                                                 |
| Vapor Pressure                                             | 0.404 bar (5.85 psig)                                                 |
| Relative dielectric strength, 1 atm, (N <sub>2</sub> =1.0) | 2.3                                                                   |
| Ozone Depletion Potential (ODP) <sup>1</sup>               | 0.0                                                                   |
| Global Warming Potential (GWP) <sup>2</sup>                | ≤1                                                                    |
| Atmospheric Lifetime                                       | 0.019 years                                                           |

Notes:

1. World Meteorological Organization (WHO) 1998, Model-derived method.
2. Intergovernmental Panel on Climate Change (IPCC) 2013 Method, 100 Year ITH.

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